# Science Skills Unit Notes

Name:

DO NOT LOSE!

Area of Focus: Lab Safety

- Handle everything as if it's pathogenic.
  - Pathogenic means that what you are handling could be an infective agent that could cause disease.
  - Clean work station periodically with proper disinfectant.
- Do not breathe vapors or put anything close to your nose to smell unless instructed.
  - When smelling, do not hold smell below nose. Make a pass from one side to the other.
- Avoid blood and other bodily fluids.
  - If you are bleeding then please contact teacher immediately to get wound cleaned and covered.
- Please check glassware for cracks or chips prior to use.
  - If glassware is broken please contact teacher.
  - Please be safe with glassware to avoid dropping and breaking. Clean immediately.
- Clean spills from the outside in.
  - Apply paper towels over the spill, then, carefully starting from the outside, wipe in.
- Please do not eat food or drink in the classroom.
  - No gum
  - Cough drops
  - Or putting strange things in your mouth.

Keep flammable solutions away from flame.

- If you have long hair then please arrange it so that it will not hang down and catch on fire.
- Know where the fire extinguisher is and how to use it.
  - We have a Carbon Dioxide all purpose fire extinguishers.
    - Find key.
    - Pull it out. (Stand back)
    - Pull handle / trigger.
    - Point at the fire until extinguished.

- Keep electrical equipment away from water and vice versa.
- Use proper safety protection.
  - Goggles covering eyes.
  - Gloves (Non-latex) for allergy reasons.
- Know where the eyewash station is and how to use it. Where is the station?
  - If you get something in your eye
    - · Get it out now!
    - · Hold eyelid open.
    - · Gently run water over your eyes
    - · Go to school nurse immediately.
- Clean glassware before and after use to avoid harmful residue.
- Avoid cutting yourself if we are using sharp objects.
  - Never cut toward yourself or others.
  - A pencil and other pointed objects can be very dangerous.
- Use common sense at all times.
  - No horseplay.
  - · No pushing.
  - No running.
  - No squirting with droppers.

Area of Focus: Magnification

Magnification: The act of expanding something in apparent size.

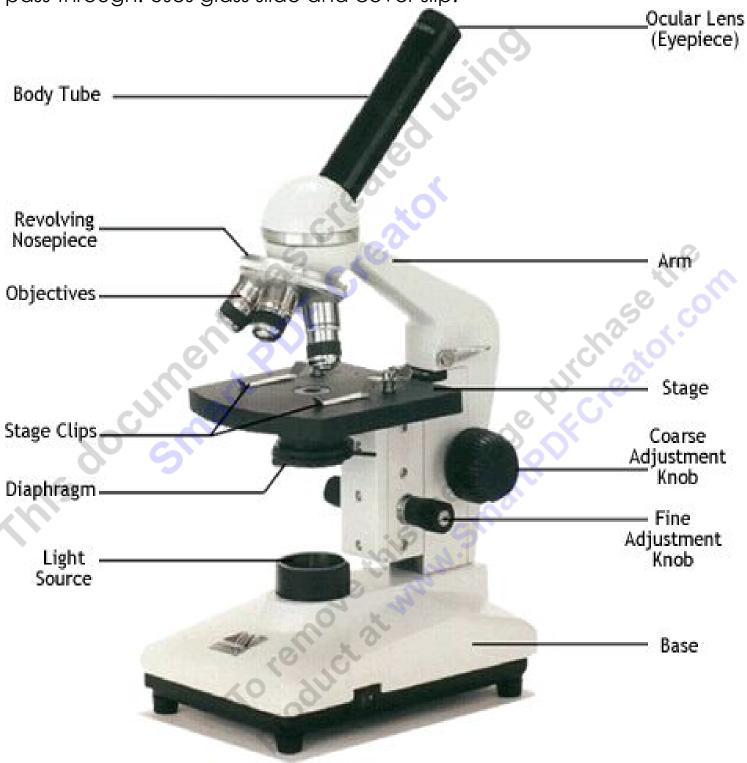
The object does not change in size.

De-magnification: To make something smaller in appearance.

This is a stereoscopic microscope. It looks at things in which light cannot pass like a bumble bee. Lets you see the image in 3D.



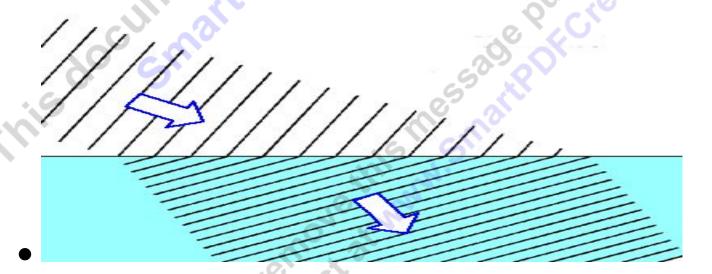
This is a light microscope. It lets you magnify images that light can pass through. Uses glass slide and cover slip.



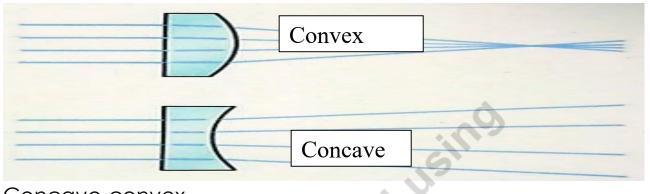
This is an electron microscope. It can magnify specimens much smaller than a light, or stereoscope, but doesn't usually view live cells or specimens

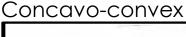


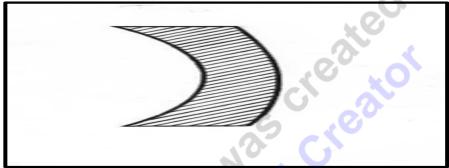
- Light is a particle and a wave and goes out in a straight line unless it bumps something.
- **Refraction:** The bending of a wave when it enters a medium where it's speed is different.



- **Diffraction**: Bending of waves.
- Lens: A transparent optical device used to converge or diverge transmitted light and to form images.
- Convex top / Concave bottom





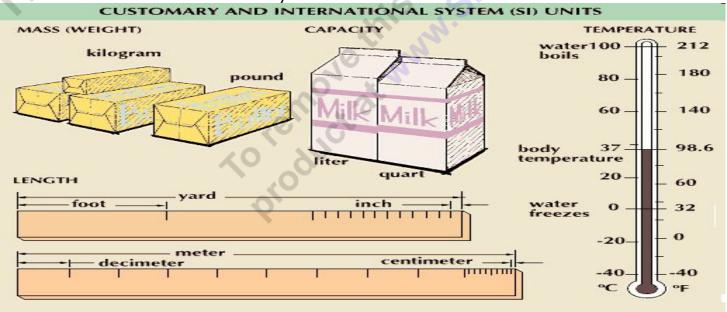


When carrying a microscope, carry it by the arm, and have one hand under the base.

Always lower the stage after use so the gears are not strained.

- Remove any slide as well.
- The finely tuned gears are what make microscopes expensive.

Area of Focus: The Metric System.



The international System of Units (SI) also known as the metric system.

Quantity	Base Unit	Symbol
Length	meter	m
Mass	kilogram	kg
Time	second	S
Electric Current	ampere	Α
Temperature	Kelvin	K
Light intensity	candela	cd
Amount of substance	mole	mol

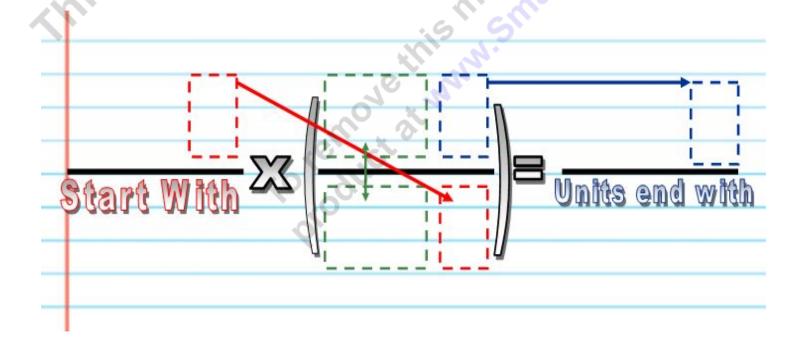
The Metric System: A measurement system based on the powers of ten.

## Converting Units

Write the conversion as a fraction

Multiply

Cancel units from the top and bottom



Scientific notation: A method for expressing, and working with, very large or very small numbers.

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5.7 \times 10^6 = 5700000
```

```
= 10^{24}
yotta [Y] 1 000 000 000 000 000 000 000 000
zetta [Z] 1 000 000 000 000 000 000 000
                                                  = 10^21
      [E] 1 000 000 000 000 000 000
                                                  = 10^18
peta [P] 1 000 000 000 000 000
                                                  = 10^15
                                                  = 10^12
tera [T] 1 000 000 000 000
     [6] 1 000 000 000
                                             (a thousand millions = a billion)
giga
mega [M] 1 000 000
                                             (a million)
                                             (a thousand)
kilo
      [k] 1 000
hecto [h] 100
                                             (a hundred)
      [da] 10
                                             (ten)
          1
                                             (a tenth)
      [d] 0.1
deci
centi [c] 0.01
                                             (a hundredth)
milli [m] 0.001
                                             (a thousandth)
micro [µ] 0.000 001
                                             (a millionth)
nano [n] 0.000 000 001
                                             (a thousand millionth)
pico [p] 0.000 000 000 001
                                                  = 10^{-12}
femto [f] 0.000 000 000 000 001
                                                  = 10^-15
atto [a] 0.000 000 000 000 000 001
                                                  = 10^-18
zepto [z] 0.000 000 000 000 000 000 001
                                                  = 10^{-21}
yocto [y] 0.000 000 000 000 000 000 001
                                                 = 10^{-24}
```

•	"I am	1828.80 mm	tall.	"
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- 182.80 cm tall." "I am
- "I am 1.8280 meters tall."
- "I am .001828 km tall."

<ul><li>King -</li></ul>	Kilometer	1000m	10³
<ul><li>Henry -</li></ul>	Hectometer	100m	10 <sup>2</sup>
<ul><li>Died -</li></ul>	Decameter	10m	101
<ul><li>While -</li></ul>	Standard	1m	100
<ul><li>Drinking -</li></ul>	Decimeter	.1m	10-1
<ul><li>Chocolate -</li></ul>	Centimeter	.01m	10-2
<ul><li>Milk -</li></ul>	Millimeter	.001m	10-³

<ul><li>Chocolate - Ce</li></ul>	entimeter	.01m	10-2
<ul><li>Milk - Mil</li></ul>	llimeter	.001m	10-3
	6	20	0,
	3.00		11/2
<ul> <li>Quantity</li> <li>Ba</li> </ul>	se Unit Syr	<u>nbol</u>	Se coll
• Length	Meter	M	No of
<ul><li>Mass</li></ul>	Kilogram	kg	To. Mo
<ul><li>Temperature</li></ul>	Kelvin	K	00,10
• Time	Second	S	SO.
<ul><li>Amount</li></ul>	Mol	mol	
<ul><li>Force</li></ul>	Newton	N SS X	
<ul> <li>Electric Current</li> </ul>	Ampere	a	
<ul><li>Luminous Intensit</li></ul>	y Candela	cd	
<ul><li>Volume</li></ul>	Liter		
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ass: The amount of avity. On earth, ma		-	as to do with
,			

Area of Focus: Mass

Metric Ton: A cubic meter filled with water or 1,000 kilograms.

The standard unit of mass in the metric system is the gram.

1 milligram = 0.001 grams 1 centigram = 0.01 grams 1 decigram = 0.1 grams 1 kilogram = 1000. grams Area of Focus: Volume, Liter, I

Volume: The three-dimensional space an object occupies.

The standard unit of volume in the metric system is the liter.

1 milliliter = 0.001 liter
 1 centiliter = 0.01 liter
 1 deciliter = 0.1 liter
 1 kiloliter = 1000. liters

Volume is also the space that matter occupies.

Matter is anything that has mass and takes up space.

How to find the volume of a cube?

Length x Width x Height - \_\_\_\_cm³

Volume of a cylinder: Where Pi = 3.14

Density: How much mass is contained in a given volume. We use grams/cm 3(grams per cubic centimeter)

Density – mass divided volume

Mass

• D = ----- grams/cm<sup>3</sup>

Volume

An object will float in water.

- Density of less than one = float.
- Density of more than one = sink.

New Area of Focus: Temperature.

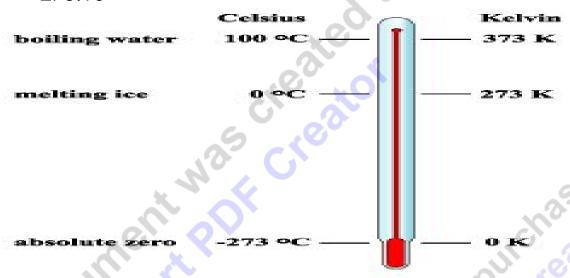
Temperature: The degree of hotness or coldness of a body or environment.

Corresponds to its molecular activity.

Temperature:

Measured in degrees Celsius.

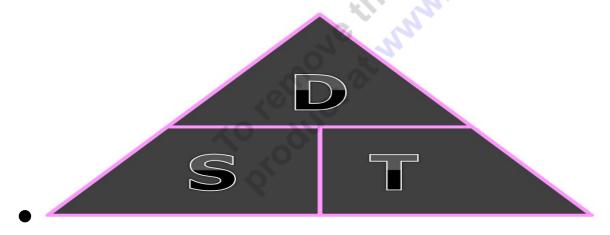
- Zero Degrees Celsius is freezing point of water, 100 degrees Celsius is boiling point.
- Kelvin Scale: Zero Kelvin is absolute zero where molecular motion stops. That is the coldest something can be. (never been reached.)
  - Water freezes at 273.16K; water boils at 373.16K. K = C + 273.16°



New Area of Focus: Time.

Time: A measuring system used to sequence events, to compare the durations of events and the intervals between them, and to quantify the motions of objects.

Speed: A measure of motion, = distance divided by time. D/T



Distance = Speed \* time (Multiply) Speed = Distance divided by time

- Velocity = Speed (distance / time) and direction.
- velocity = Distance Divided by Time
- Acceleration = The rate of change in velocity. (m/s)
  The final velocity the starting velocity, divided by time.

also... 
$$a = (v_2 - v_1)/(t_2 - t_1)$$

The SI Unit for acceleration is m/s<sup>2</sup>

- Deceleration To slow velocity.

  The same formula but value will be negative. –m/s²
- Momentum: A measure of the motion of a body equal to the product of its mass and velocity.
  - Momentum = Mass times velocity
  - Law Conservation of Momentum: The momentum of an object is the product of its mass and its velocity.
  - Angular momentum: Rotating objects tend to remain rotating at the same speed / direction unless acted upon.
  - When you draw the weights inward, your moment of inertia decreases, and your velocity increases (spin faster).
- Amount of Work (w) done depends on two things:
   The amount of Force (F) exerted.

The Distance (d) over which the Force is applied.

- Equation for Work  $w = F \times d$
- Joule: Unit of energy, work, or amount of heat.
  - Equal to the energy expended in applying a force of one newton through a distance of one meter.

New Area of Focus: Some of the other SI units.

The mole: The molecular weight of a substance expressed in grams.

Ampere: The unit of measurement of electric current, equal to one coulomb per second.

Coulomb: The measurement of a number of electrons.

Candela: The unit of luminous intensity. One candela is equivalent to 12.57 lumens.

Use to be the light of a standard candle.

age Phichase the age to reat or son New Area of Focus: Observation, Inferences, and the Scientific Method.

#### Science is...

A study of natural phenomenon.

A systematic study and method.

Knowledge through experience.

### A good Scientist is....

- Is safe!
- Is accurate, precise and methodical.
- Is unbiased, a seeker of the truth.
- Can observe and question.
- Can find solutions, reasons, and research.
- Works in all weather conditions if safe.
- Can overcome obstacles.
- Collaborates (talks) with others.

Science is a systematic attempt to get around human limitations.

 Science tries to remove personal experience from the scientific process.

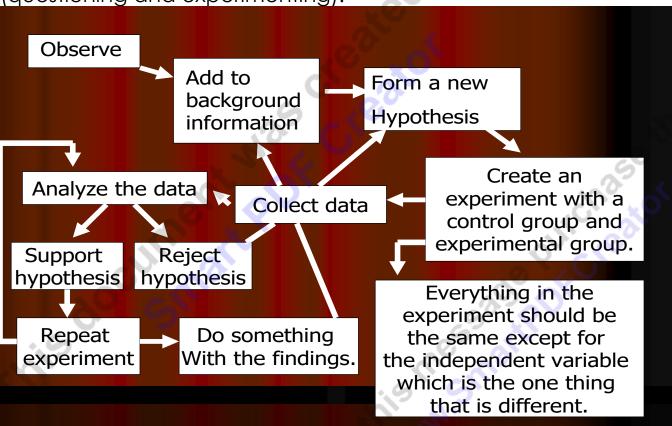
#### TRY AND WRITE WITHOUT PERSONAL PRONOUNS.

 DO NOT USE...I, me, you, he, she, we, you, they, them, theirs, names, etc

#### Types of scientists...

- Biology The study of life.
- Geology The study of earth.
- Chemistry The study of Matter.
- Physics The study of matter and energy.
- Many more...

Scientific method: A process that is the basis for scientific inquiry (questioning and experimenting).



Variable: Changing quantity of something.

- Independent: (Change) The variable you have control over, what you can choose and manipulate.
- Dependent: (Observe) What you measure in the experiment and what is affected during the experiment.
- Control: (Same) Quantities that a scientist wants to remain constant so it a fair test.

Observation – Anything you can see, hear, smell, touch, taste, (Using your senses).

Inference: A conclusion based on your observations.

Hypothesis: An educated guess to your problem / question that is J NOTLO testable.

HOLD ON TO THESE NOTES. DO NOT LOSE! TO REPROVE ALTINUM. STRAITE PROCESSAGE PRINCIPAGE PROCESSAGE PRINCIPAGE PROCESSAGE PRINCIPAGE PROCESSAGE PRINCIPAGE PROCESSAGE PRINCIPAGE PRINCIPAGE PROCESSAGE PRINCIPAGE PROCESSAGE PRINCIPAGE PRINC

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