# Science Skills Unit Notes

Name:

DO NOT LOSE!

Area of Focus: Lab Safety

•	Handle everything as if it's
	<ul> <li>Pathogenic means that what you are handling</li> </ul>
	could be an infective agent that could cause
	<ul> <li>— work station periodically with</li> </ul>
	proper disinfectant.
•	Do not vapors or put anything
	close to your nose to smell unless instructed.
	<ul> <li>When smelling, do not hold smell below nose.</li> </ul>
	Make a from one side to the other.
•	Avoid and other bodily fluids.
	<ul> <li>If you are bleeding then please contact</li> </ul>
•	immediately to get wound
	cleaned and covered.
•	Please check for cracks or chips
	prior to use.
	<ul> <li>If glassware is broken please contact</li> </ul>
	<ul> <li>Please be safe with glassware to avoid</li> </ul>
	dropping and breaking. Clean immediately.
•	Clean spills from the in.
	<ul> <li>Apply paper towels over the spill, then, carefully</li> </ul>
	starting from the outside wipe in

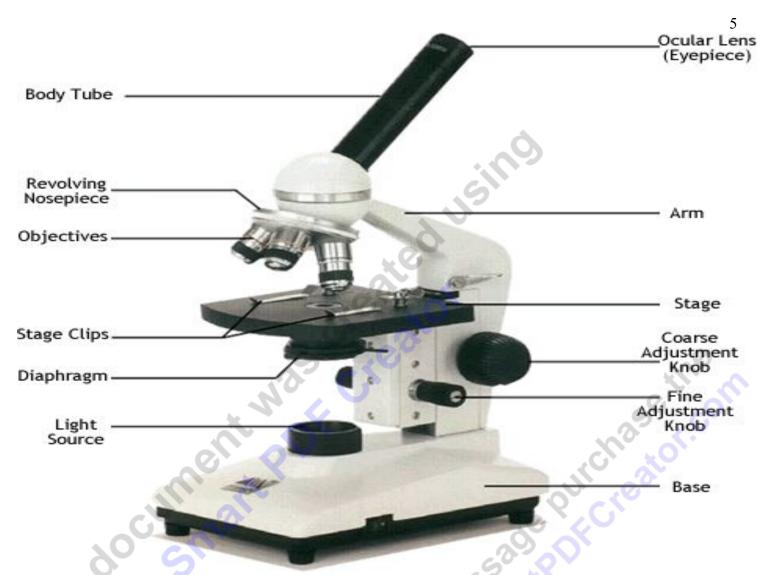
<ul> <li>Please do</li> </ul>	eat tooa or ar	ink in the
classroom.		
<ul><li>No gum</li></ul>		
<ul><li>Cough drops</li></ul>	S	, Ó)
-	range things in yo	our mouth.
Keep		ons away from
flame.		ons array nom
<ul> <li>If you have long</li> </ul>	ther	n please arrange it
so that it will not		
<ul> <li>Know where the</li> </ul>		
• •	ille exili igusi lei	is and now to use
it.	Sarban	
We have a C		all purpose
fire extinguisl		, ch to
<ul> <li>Find key.</li> </ul>		July 16 yr
	. (Stand back)	'e SC
	dle / trigger.	30 O
• Point at t	he fire until extin	
<ul> <li>Keep</li> </ul>		oment away from
water and vice	versa.	
<ul><li>Use proper s</li></ul>	fl. "h.	_ protection.
• G	covering	eyes.
• G	(Non-latex)	for allergy reasons.
<ul> <li>Know where the</li> </ul>	(e, 'C,	station is
and how to use	it. Where is the s	tation?
<ul> <li>If you get so;</li> </ul>	mething in your_	
<ul> <li>Get it ou</li> </ul>	t now!	
<ul> <li>Hold eye</li> </ul>	lid open.	
<ul> <li>Gently ru</li> </ul>	ın water over you	ur eyes

<ul> <li>Go to school nurse immediately.</li> </ul>
<ul> <li>Clean before and after use</li> </ul>
to avoid harmful residue.
<ul> <li>Avoid yourself if we are using</li> </ul>
sharp objects.
<ul> <li>Never cut toward yourself or others.</li> </ul>
<ul> <li>A pencil and other pointed objects can be ver</li> </ul>
dangerous.
<ul> <li>Use common at all times.</li> </ul>
No horseplay.
No pushing.
No running.
<ul> <li>No squirting with droppers.</li> </ul>
Area of Focus: Magnification
Area or rocus. Magrimcanon
Magnification: The act of
something in apparent size.
<ul> <li>The object does not change in size.</li> </ul>
De-magnification: To make something
in appearance.
C. C.

This is a stereoscopic microscope. It looks at things in which light \_\_\_\_\_ pass like a bumble bee. Lets you see the image in 3D.



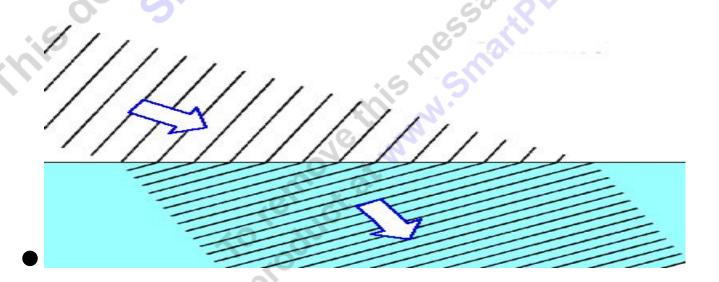
This is a light microscope. It lets you magnify images that light can pass through. Uses \_\_\_\_\_\_ slide and cover slip.



This is an \_\_\_\_\_\_ microscope. It can magnify specimens much smaller than a light, or stereoscope, but doesn't usually view live cells or specimens



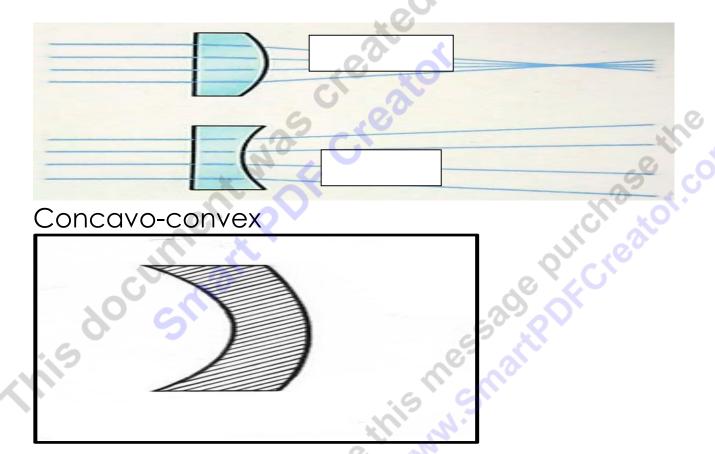
- Light is a \_\_\_\_\_\_ and a wave and goes out in a straight line unless it bumps something.
- Refraction: The \_\_\_\_\_\_ of a wave when it enters a medium where it's speed is different.



■ Diffraction: Bending of \_\_\_\_\_

■ Lens: A transparent optical device used to c\_\_\_\_\_ or diverge transmitted light and to form images.

### ■ Convex top / Concave bottom



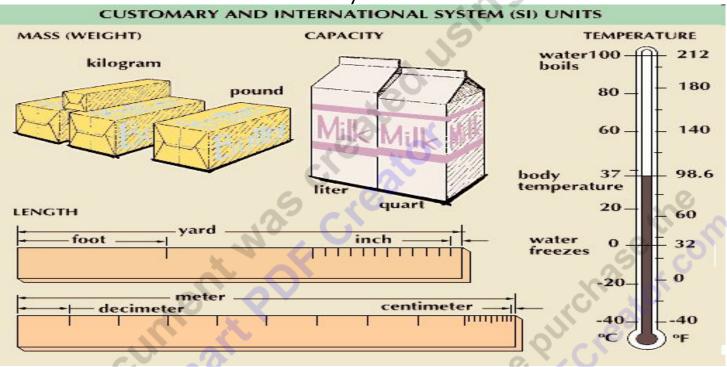
When carrying a microscope, carry it by the \_\_\_\_\_, and have one \_\_\_\_\_ under the base.

Always \_\_\_\_\_ the stage after use so the gears are not strained.

- Remove any slide as well.

 The finely tuned gears are what make microscopes expensive.

Area of Focus: The Metric System.

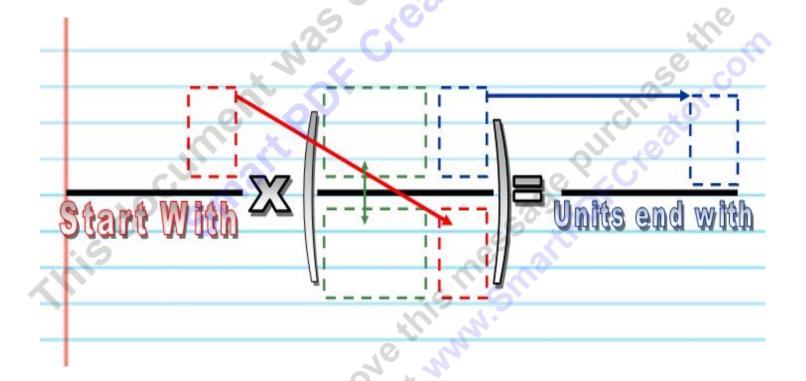


The international System of U\_\_\_\_\_ (SI) also known as the metric system.

Quantity	Base Unit	Symbol
Length	meter	m
Mass	kilogram	kg
Time	second	s
Electric Current	ampere	Α
Temperature	Kelvin	K
Light intensity	candela	cd
Amount of substance	mole	mol

The Metric System: A measurement system based on the powers of \_\_\_\_\_.

Converting Units
Write the conversion as a fraction
Multiply
Cancel units from the top and bottom



Scientific notation: A method for expressing, and working with, very \_\_\_\_\_ or very \_\_\_\_ numbers.

# $5.7 \times 10^6 = 5700000$

123456

```
yotta [Y] 1 000 000 000 000 000 000 000 000
                                                 = 10^24
zetta [Z] 1 000 000 000 000 000 000 000
                                                 = 10^21
exa [E] 1 000 000 000 000 000 000
                                                 = 10^18
peta [P] 1 000 000 000 000 000
                                                 = 10^15
tera [T] 1 000 000 000 000
                                                 = 10^12
     [G] 1 000 000 000
                                            (a thousand millions = a billion)
giga
mega [M] 1 000 000
                                            (a million)
     [k] 1 000
                                            (a thousand)
kilo
                                            (a hundred)
hecto [h] 100
                                            (ten)
deca
      [da] 10
          1
deci [d] 0.1
                                            (a tenth)
                                            (a hundredth)
centi [c] 0.01
                                            (a thousandth)
milli [m] 0.001
micro [µ] 0.000 001
                                            (a millionth)
nano [n] 0.000 000 001
                                            (a thousand millionth)
pico [p] 0.000 000 000 001
                                                 = 10^-12
femto [f] 0.000 000 000 000 001
                                                 = 10^-15
atto [a] 0.000 000 000 000 000 001
                                                 = 10^{-18}
zepto [z] 0.000 000 000 000 000 000 001
                                                 = 10^{-21}
yocto [y] 0.000 000 000 000 000 000 000 001
```

<ul><li> "I am 182.80 d</li><li> "I am 1.828</li></ul>	mm tall." cm tall." 0 meters tall." 828 km tall."	0	
<ul> <li>King -</li> <li>Henry -</li> <li>Died -</li> <li>While -</li> <li>Stand</li> <li>Drinking -</li> <li>Chocolate</li> <li>Milk -</li> </ul>	meter meter dard meter meter meter	1000m 100m 10m 1m .1m .01m .001m	10 <sup>3</sup> 10 <sup>2</sup> 10 <sup>1</sup> 100 10- <sup>1</sup> 10- <sup>2</sup>
<ul> <li>Quantity</li> <li>Length</li> <li>M</li> <li>Temperature</li> <li>T</li> <li>Amount</li> <li>Force</li> <li>Electric Current</li> <li>Luminous Intensi</li> <li>V</li> </ul>	Base Unit  Kilogram  K Second  M Newton  A ty Candela Liter	Sym M K s mol N a cd I	bol_

Area of Focus: Mass

Mass: The amount of In an object. Weight has to do with gravity. On earth, mass and weight are the same.
Metric Ton: A meter filled with water or 1,000 kilograms.
The standard unit of in the metric system is the gram.  1 milligram = 0.001 grams  1 centigram = 0.01 grams  1 decigram = 0.1 grams  1 kilogram = 1000. grams
Area of Focus: Volume, Liter, I
Volume: The three-dimensional an object occupies.
The standard unit of in the metric system is the liter.  • 1 milliliter = 0.001 liter  1 centiliter = 0.1 liter  1 deciliter = 0.1 liter  1 kiloliter = 1000. liters

Volume is also the	that matter
occupies.	
<ul><li>Matter is anything that h</li></ul>	nas and takes
UP	
•	
How to find the volume of a c	cube?
<ul><li>Length x Width x Height</li></ul>	
olume of a cylinder: Where Pi	= 3.14
	"We
ensity: How much	is contained in a
iven volume. We use grams/c	
entimeter)	(9.5)
<ul><li>Density – mass</li></ul>	volume
Mass	60.0
	ms/cm³
Volume	1137 6111
VOIOTTIC	M.

An object will float in water.

- Density of less than one = \_\_\_\_\_\_.
- Density of more than one = \_\_\_\_\_\_.

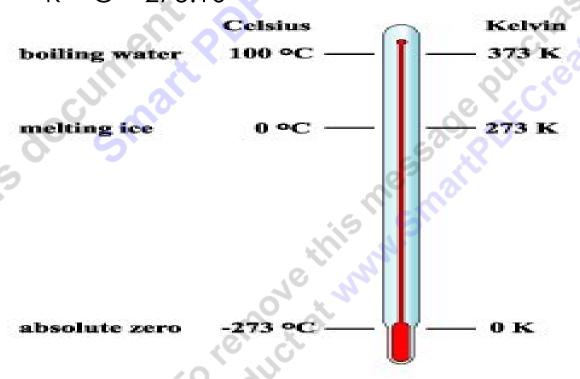
New Area of Focus: Temperature.

Temperature: The degree of hotness or coldness of a body or environment.

Corresponds to its molecular activity.

### Temperature:

- Measured in degrees \_\_\_\_\_\_\_.
- Zero Degrees Celsius is freezing point of water,
   100 degrees Celsius is boiling point.
- Kelvin Scale: Zero Kelvin is absolute \_\_\_\_\_\_\_
   where molecular motion stops. That is the coldest something can be. (never been reached.)
  - Water freezes at 273.16K; water boils at 373.16K.
     K = C + 273.16°

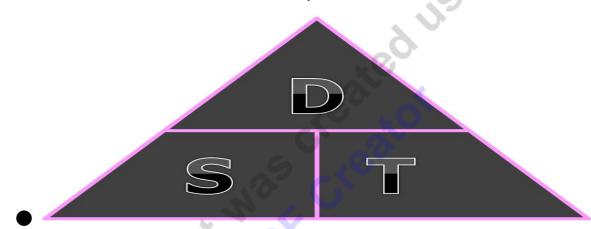


New Area of Focus: Time.

Time: A measuring system used to sequence \_\_\_\_\_, to compare the durations of events

and the intervals between them, and to quantify the motions of objects.

Speed: A measure of motion, = distanceby time. D/T



Distance = Speed \* time (M\_\_\_\_\_)
Speed = Distance divided by time
Time = Distance \_\_\_\_\_ by Speed

- Velocity = Speed (distance / time) and
- velocity = Distance Divided by Time
- Acceleration = The rate of \_\_\_\_\_ in velocity. (m/s)

The final velocity – the \_\_\_\_\_ velocity, divided by time.

also... 
$$a = (v_2 - v_1)/(t_2 - t_1)$$

The SI Unit for acceleration is m/s<sup>2</sup>

■ Deceleration – To	velocity.
The same formula but y	value will be negative. –
m/s <sup>2</sup>	usill'
■ Momentum: A measure of	the of
a body equal to the produ	ict of its mass and
velocity.	*O,
■ Momentum = Mass	velocity
Law Conservation of M	lomentum: The
momentum of an obje	ct is the product of its
mass and its	- Charlet
■ Angular momentum: R	otating objects tend to
remain rotating at the	same speed / direction
unless acted upon.	20 OF
■ When you draw the we	
moment of inertia dec	reases, and your velocity
increases (spin faster).	5 5
	"MA"
<ul><li>Amount of Work (w) done</li></ul>	
The amount of F	(F) exerted.
The Distance (d) over v	which the Force is
pplied.	
■ Equation for Work - $w = F$	

a

<ul><li>Joule: Unit of</li></ul>	, work, or amount
of heat.	
<ul> <li>Equal to the energy force of one newtor</li> </ul>	expended in applying a through a
of one meter.	
	ted us
New Area of Focus: Some o	f the other SI units.
The mole: The molecular	of a
substance expressed in grai	
	cha of
Ampere: The unit of measur	
current, equal to one coulo	
Coulomb. The meas	urement of a number of
	CS WILL
Candela: The unit of	intensity. One
candela is equivalent to 12.	57 lumens.
<ul><li>Use to be the light o</li></ul>	f a standard
New Area of Focus: Observe	ation, Interences, and the
Scientific Method.	
Science is	
A study of natural	
A systematic study and	•

Knowledge through experience.

A good Scientist is			
• Is	<u>!</u>	20	
<ul><li>Is accurate, pr</li></ul>	recise and		
● Is unbiased, a:	seeker of the	truth.	
<ul><li>Can</li></ul>	an	d question.	
<ul> <li>Can find solution</li> <li>research.</li> </ul>	ons,	, and	k
<ul><li>Works in all we</li></ul>	eather condition	ons if safe.	O.
<ul><li>Can overcome</li></ul>	e obstacles.		1/1/20
<ul><li>Collaborates (</li></ul>	) \	with others.	3 6

Science is a systematic attempt to get around human limitations.

 Science tries to remove personal experience from the scientific process.

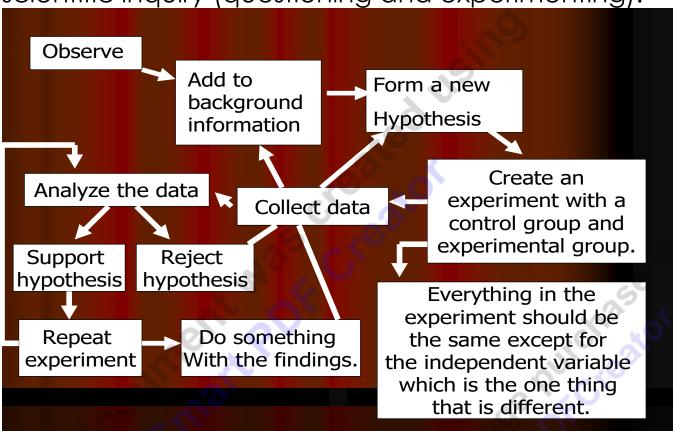
### TRY AND WRITE WITHOUT PERSONAL PRONOUNS.

● DO NOT USE...I, me, you, he, she, we, you, they, them, theirs, names, etc

## Types of scientists...

- Biology The study of life.
- Geology The study of earth.
- Chemistry The study of Matter.
- Physics The study of matter and energy.
- Many more...

Scientific method: A process that is the basis for scientific inquiry (questioning and experimenting).



Variable: Changing quantity of something.

- Independent: (\_\_\_\_\_\_\_\_) The variable you have control over, what you can choose and manipulate.
- Dependent: (\_\_\_\_\_\_\_) What you measure in the experiment and what is affected during the experiment.
- Control: (\_\_\_\_\_\_\_) Quantities that a scientist wants to remain constant so it a fair test.

Observation – Anything you can see, hear, smell, touch taste, (Using your).
Inference: A based on your observations.
Hypothesis: An educatedto your problem / question that is testable.
HOLD ON TO THESE NOTES. DO NOT LOSE!  Copyright © 2010 Ryan P. Murphy